

Objective 1 Evaluate common logarithms using a calculator.









CLASSROOM EXAMPLE 3	Finding Hydronium Ion Concentration
Find the hydroniu	im ion concentration of a solution with pH 4.6.
Solution:	
	$pH = -\log \left[H_3O^+ \right]$
	$4.6 = -\log[H_3O^+]$
	$-4.6 = \log[H_{3}O^{+}]$
	$-4.6 = \log_{10}[H_3O^+]$
	$10^{-4.6} = [H_3O^+]$
	$2.5 \times 10^{-5} \approx [H_3O^+]$



Decibel Level	Example
60	Normal conversation
90	Rush hour traffic, lawn mower
100	Garbage truck, chain saw, pneumatic dril
120	Rock concert, thunderclap
140	Gunshot blast, jet engine
180	Rock launching pad



Evaluate natural logarithms using a calculator.

Logarithms used in applications are often **natural logarithms**, which have as base the number e.

The number e, (\approx 2.718281828), like π , is a **universal constant.** Since it is an irrational number, its decimal expansion never terminates and never repeats.

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Objective 4

Use natural logarithms in applications.

 CLASSROOM EXAMPLE 6
 Applying a Natural Logarithm Function

 The altitude in meters that corresponds to an atmospheric pressure of x millibars is given by the logarithmic function defined by f(x) = 51,600 - 7457 ln x. Approximate the altitude at 700 millibars of pressure.

 Solution:
 $f(x) = 51,600 - 7457 \ln x$ $f(700) = 51,600 - 7457 \ln 700$ $\approx 2748.6 \approx 2700$

 Atmospheric pressure is 700 millibars at approximately 2700 m.



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